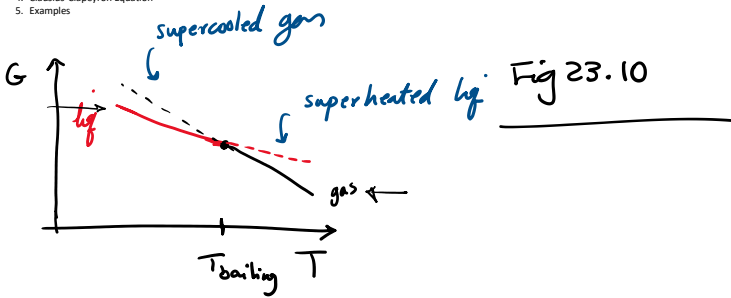


1. Phase Equilibria => Only consider one substance at different phases
2. G-T and P-T phase diagrams
3. Clausius Equation
4. Clausius-Clapeyron Equation
5. Examples

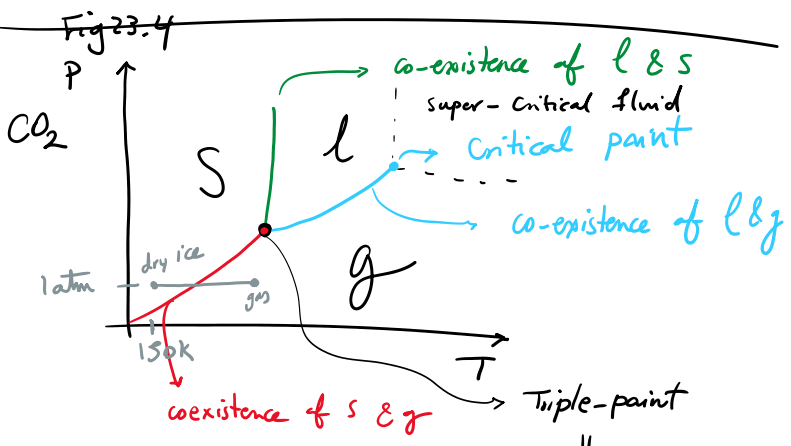


100°C = 373K at 1 atm

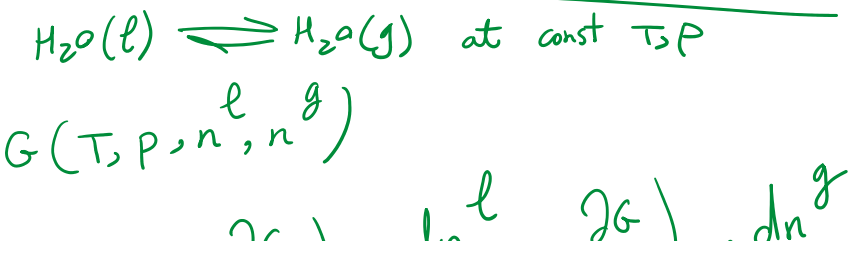
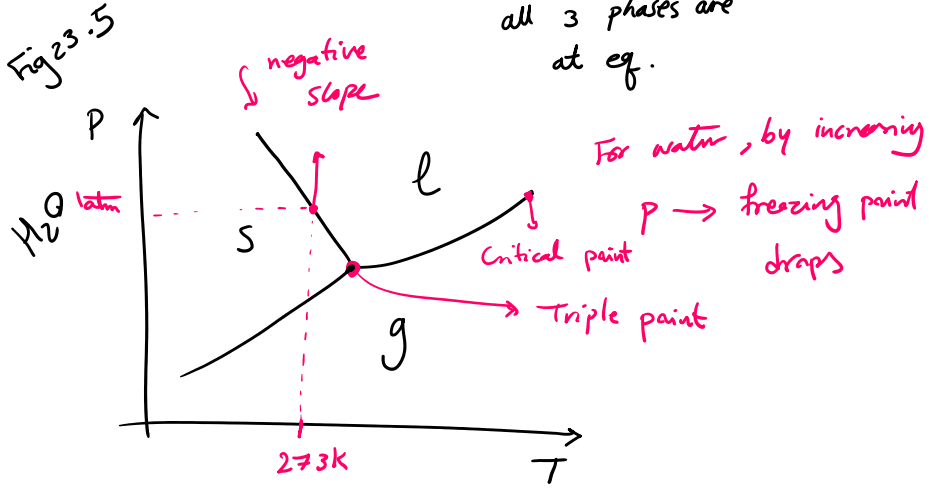
$$\text{slope} = \frac{dG}{dT} = -S$$

$$dG = \cancel{V}dp - SdT$$

$S_{\text{gas}} \gg S_{\text{liq}}$



all 3 phases are at eq.



$G(T, P, \dots)$

$$dG = \left(\frac{\partial G}{\partial n^l} \right)_{P, T} \cdot dn^l + \left(\frac{\partial G}{\partial n^g} \right)_{P, T} \cdot dn^g$$

$$dG = \mu^l \cdot \underline{dn^l} + \mu^g \cdot dn^g$$

$$\underline{dn^l} = - \underline{dn^g} \quad \text{Conservation of matter}$$

$$dG = \mu^l \cdot (- \underline{dn^g}) + \mu^g \cdot \underline{dn^g}$$

$$dG = (\mu^g - \mu^l) \cdot dn^g$$

↪ at eq: $dG = 0 \rightsquigarrow \mu^g - \mu^l = 0$

$$\boxed{\mu^g = \mu^l}$$

$$\alpha \rightleftharpoons \beta \quad G(T, P, n^\alpha, n^\beta)$$

$$dG = \left(\frac{\partial G}{\partial n^\alpha} \right)_{T, P} \cdot dn^\alpha + \left(\frac{\partial G}{\partial n^\beta} \right)_{T, P} \cdot dn^\beta$$

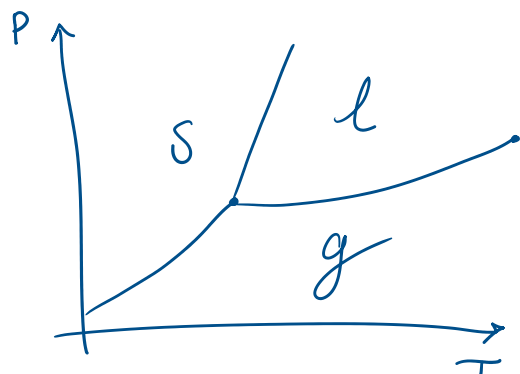
$$dG = \mu^\alpha \cdot dn^\alpha + \mu^\beta \cdot dn^\beta$$
$$dn^\alpha = -dn^\beta$$

$$dG = \mu^\alpha (-dn^\beta) + \mu^\beta \cdot dn^\beta$$

$$dG = (\mu^\beta - \mu^\alpha) \cdot dn^\beta \quad \left[\begin{array}{c} \beta \\ \alpha \end{array} \right]$$

$$dG = (\mu^r - \mu^\alpha) \cdot dn^i$$

at eq $dG=0 \rightarrow \mu^\beta = \mu^\alpha$



slope = $\left(\frac{dP}{dT}\right)$

At eq: $dG=0$ $\mu^\alpha = \mu^\beta$

$$\mu^\alpha(T, P) = \mu^\beta(T, P)$$

Take derivation w.r.t T, P

$$d\mu^\alpha(T, P) = d\mu^\beta(T, P)$$

$$\left(\frac{\partial \mu^\alpha}{\partial P}\right)_T \cdot dP + \left(\frac{\partial \mu^\alpha}{\partial T}\right)_P \cdot dT = \left(\frac{\partial \mu^\beta}{\partial P}\right)_T \cdot dP + \left(\frac{\partial \mu^\beta}{\partial T}\right)_P \cdot dT$$

$$\left(\frac{\partial \mu}{\partial P}\right)_T = \frac{\partial}{\partial P} \left(\frac{\partial G}{\partial n} \right)_T = \frac{\partial}{\partial n} \left(\frac{\partial G}{\partial P} \right)_T = \frac{\partial V}{\partial n} = \bar{V}$$

$$dG = V dP - S dT$$

$$\left(\frac{\partial \mu}{\partial T}\right)_P = \frac{\partial}{\partial T} \left(\frac{\partial G}{\partial n} \right)_P = \frac{\partial}{\partial n} \left(\frac{\partial G}{\partial T} \right)_P = -S$$

$$\left(\frac{\partial \mu}{\partial T}\right)_P = \frac{\partial}{\partial T} \left(\frac{\partial G}{\partial n} \right)_P = \frac{\partial}{\partial n} \left(\frac{\partial G}{\partial T} \right)_P$$

$$= -\frac{\partial S}{\partial n} = -\bar{S}$$

$$\bar{V}^\alpha dP - \bar{S}^\alpha dT = \bar{V}^\beta dP - \bar{S}^\beta dT$$

$$\bar{V}^\alpha dP - \bar{V}^\beta dP = \bar{S}^\alpha dT - \bar{S}^\beta dT$$

$$\underbrace{(\bar{V}^\alpha - \bar{V}^\beta)}_{\Delta \bar{V}_{\text{trans}}} dP = \underbrace{(\bar{S}^\alpha - \bar{S}^\beta)}_{\Delta \bar{S}_{\text{trans}}} dT$$

$$\Delta \bar{V}_{\text{trans}} \cdot dP = \Delta \bar{S}_{\text{trans}} \cdot dT$$

$$\frac{dP}{dT} = \frac{\Delta \bar{S}_{\text{trans}}}{\Delta \bar{V}_{\text{trans}}}$$

at eq.: rev. process

$$\Delta \bar{S}_{\text{trans}} = \frac{\Delta \bar{H}_{\text{trans}}}{T}$$

dP

$\Delta \bar{H}_{\text{trans}}$

$$\frac{dP}{dT} = \frac{\Delta H_{\text{trans}}}{T \Delta \bar{V}_{\text{trans}}}$$

clapeyron-eg



$$\Delta \bar{V}_{\text{trans}} = \bar{V}_l - \bar{V}_s > 0 \rightarrow \frac{dP}{dT} > 0$$

Water

$$\Delta \bar{V}_{\text{trans}} = \bar{V}_l - \bar{V}_s < 0 \rightarrow \frac{dP}{dT} < 0$$
