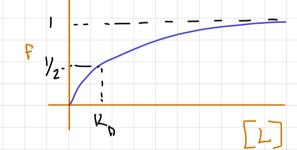
dissociation PL=P+L

dissociation
$$PL = P+L$$
 $V_b = (PL) (PL)^o + ($

with
$$L_b^* = PTLLT$$
 having units of M: moll



binding isothermal

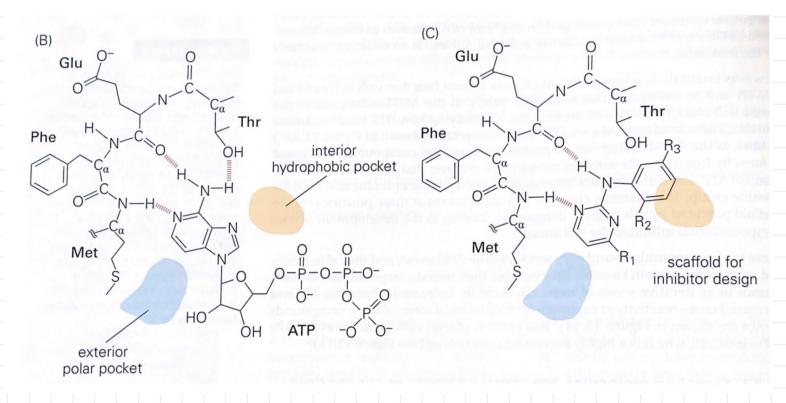
affinity is specificity,

usually KD is not so different from the natural/physiological [L]

why? If mutations lower LD, it doesn't change f much.

· can stop the "Off" state which is important for signaling,

· If mutations raise Vp, protien could fail to bind ligand



- · drug development begins by identifying proteins that are critical to disease progression, and then focuses on design or discovery of mouchles that inhibit the protein function
- ·The molecules that are first discovered to inhibit the target protein are called "lead compounds". Then these are refined.

example: imatinib > blocks Abl Linase to treat chronic mylogenous lukemia. Abl Kinase catalyzes phosporylation of specific typosine residues that control signal transmissions between cells.

By studying how Abl Kinase bonds ATP, chemists could design a scaffold to competitive inhibition

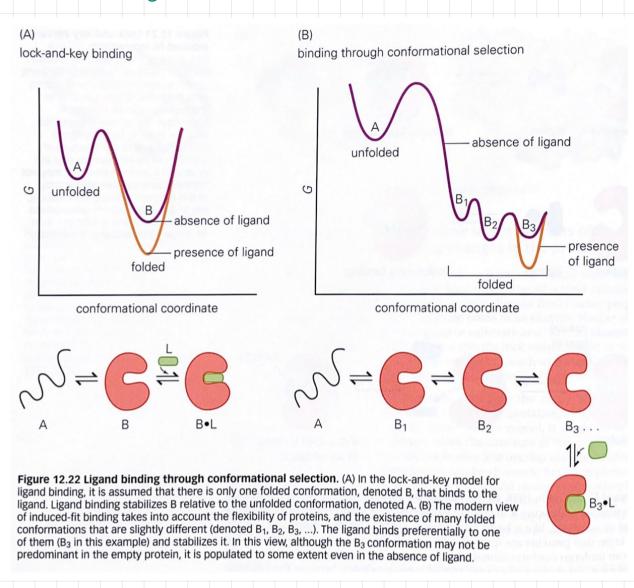
and low cross-reactivity to other kinases

Pinactive

Pactive P-imatinib

P-Dasatinib > dasatinib bonds 350 x stronger, but not as

*Protein switching



X Next class, back to Chapter 24