## Chem 3322 homework \#7, due March 22, 2024

## Problem 1 - harmonic oscillator wavefunctions

In class, we found that the stationary states of the 1 d harmonic oscillator have the form

$$
\begin{equation*}
\psi_{n}=A_{n} \times \mathrm{n}^{\mathrm{th}} \text { order polynomial } \times e^{-\alpha x^{2}} \tag{1}
\end{equation*}
$$

where $A_{n}$ is a normalization constant, and where

$$
\begin{equation*}
\alpha=\frac{m \omega}{2 \hbar} \tag{2}
\end{equation*}
$$

We did not derive a general formula for the polynomials, although we noted that each polynomial contains only even, or only odd, powers of $x$. These could, although its not very practical, be determined by orthogonality. For all of this question, express all your answers and do all your work in terms of the parameter $\alpha$ only. (You will need to use integral tables)
a) In particular, the second excited state $\psi_{2}$ has the form

$$
\begin{equation*}
\psi_{2}=A_{2}\left(x^{2}+c\right) e^{-\alpha x^{2}} \tag{3}
\end{equation*}
$$

Find the constant $c$ by requiring that $\psi_{2}$ be orthogonal to the ground state $\psi_{0}$.
b) $\psi_{2}$ is also orthogonal to the first excited state $\psi_{1}$. Why? (hint: symmetry)
c) Determine the normalization constant $A_{0}$ for the ground state.

## Problem 2 - atomic orbitals

Do problem 6-21 from your textbook. Use Table 6.5 and the Jacobian (equation D.3).

## Problem 3 - atomic orbitals

Do problem 6-30 from your textbook.

